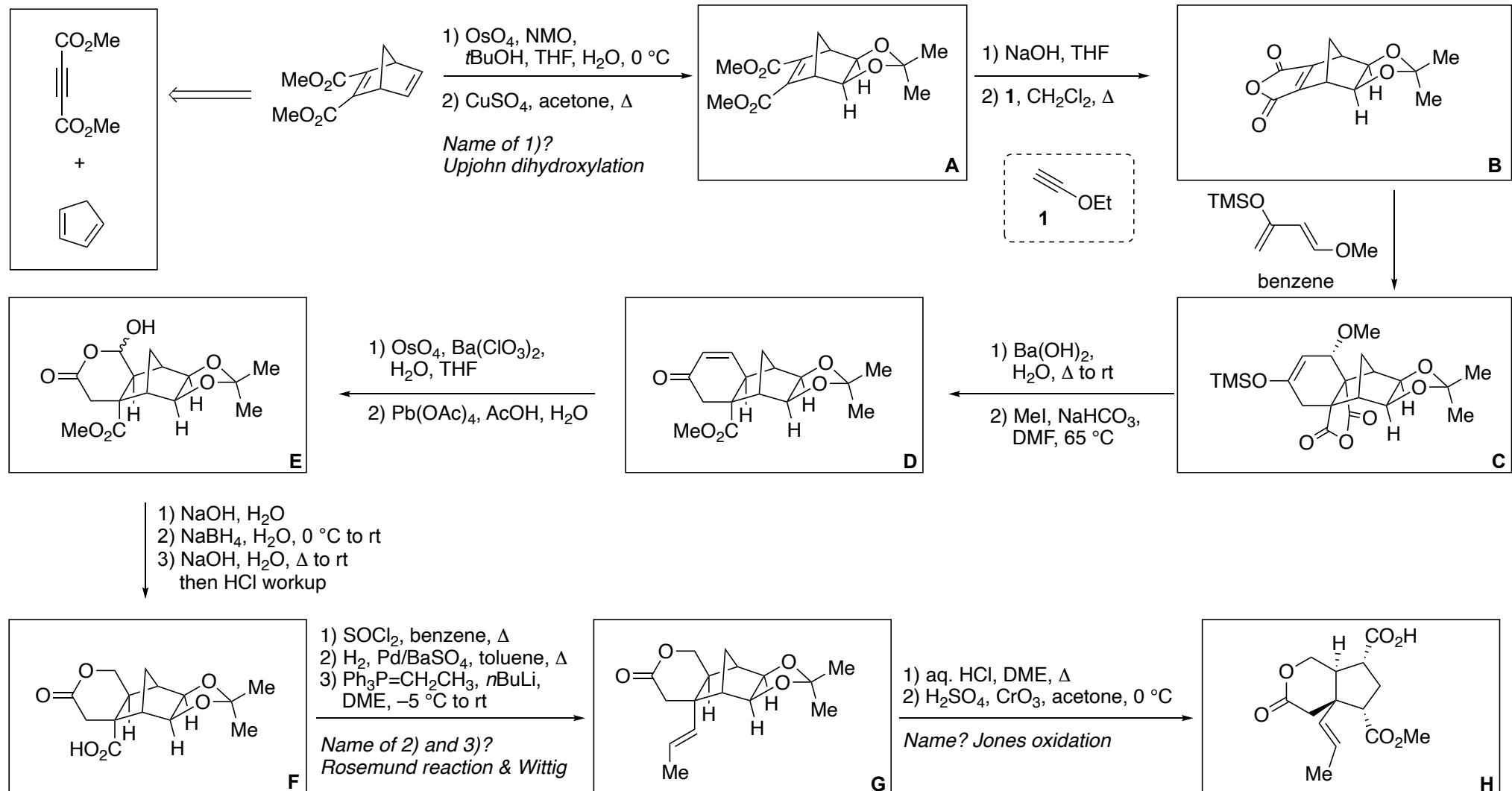
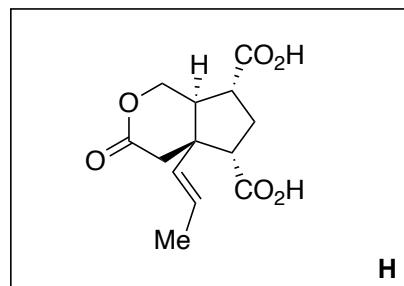


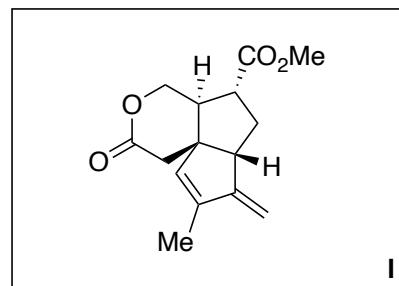


# Pentalenolactone



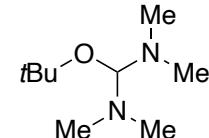
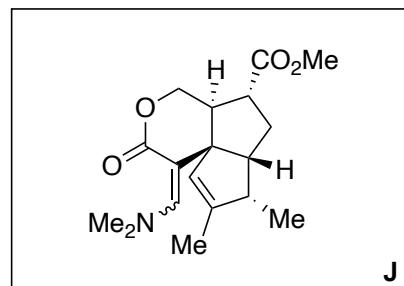


1)  $\text{H}_2\text{SO}_4$ , MeOH,  
0 to 5 °C  
2)  $\text{SOCl}_2$ , benzene,  $\Delta$   
then  $\text{AlCl}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  
0 °C to rt  
3)  $\text{BrPh}_3\text{PMe}$ , NaH,  
DMSO, 10 °C



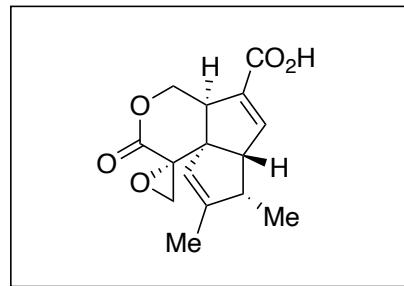
1)  $\text{H}_2$ ,  $\text{RhCl}(\text{PPh}_3)_3$ , benzene  
2) 2, rt to 96 °C

Name of 1)?  
Wilkinson hydrogenation

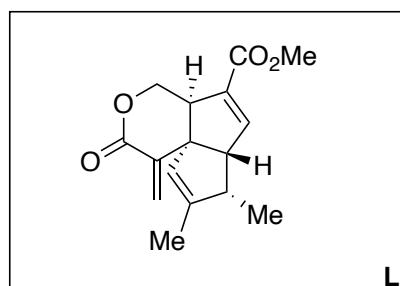


2  
Name of reagent?  
Bredereck's reagent

1)  $\text{SiO}_2$ ,  $\text{CH}_2\text{Cl}_2$ ,  $\text{H}_2\text{O}$   
2)  $\text{NaBH}_4$ , MeOH, 0 °C  
3)  $\text{MsCl}$ , pyridine,  
0 °C to rt  
4) DBU, benzene



1) DIBAL-H, DME, benzene, -78 °C  
2)  $\text{VO}(\text{acac})_2$ ,  $t\text{BuOOH}$ , benzene,  $\Delta$   
3)  $\text{H}_2\text{SO}_4$ ,  $\text{CrO}_3$ , acetone, 0 °C  
4)  $\text{KOH}$ , THF, 0 °C to rt



1) LDA, then  $\text{PhSeCl}$ ,  
THF, -78 °C  
2)  $\text{NaIO}_4$ , MeOH,  $\text{H}_2\text{O}$

